

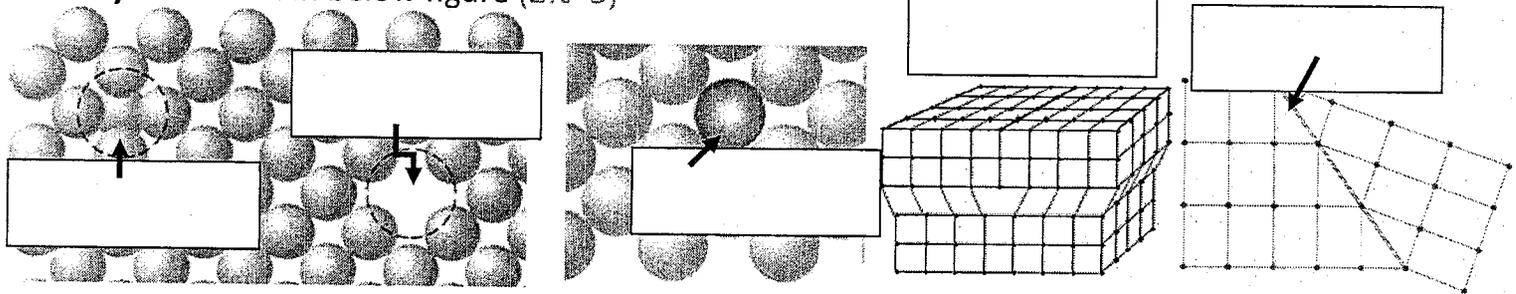
大同大學 106 學年度(寒)轉學入學考試試題

考試科目:材料導論 系別:材料工程學系 第2-1頁

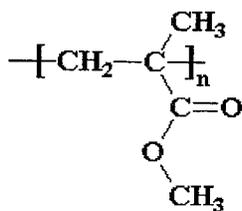
註:本次考試 不可以參考自己的書籍及筆記; 不可以使用字典; 不可以使用計算器。

1. Rank and explain the magnitudes of the diffusion coefficients from greatest to least for the following systems: (1) N in Fe at 700°C (2) Cr in Fe at 700°C (3) N in Fe at 900°C (4) Cr in Fe at 900°C. (3, 7%)
 (Note: Both Fe and Cr have the BCC crystal structure, and the atomic radii for Fe, Cr, and N are 0.124, 0.125, and 0.065 nm, respectively)

2. Identify the defect in below figure (2%*5)



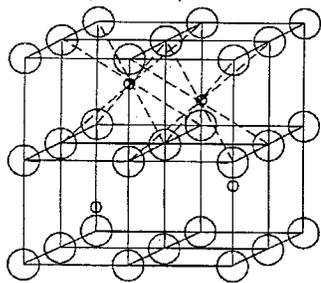
3. Is it possible to have a poly(methyl methacrylate) homopolymer with the following molecular weight data and a degree of polymerization of 530? Why or why not? (10%)



Molecular Weight Range (g/mol)	w_i	x_i
8,000–20,000	0.02	0.05
20,000–32,000	0.08	0.15
32,000–44,000	0.17	0.21
44,000–56,000	0.29	0.28
56,000–68,000	0.23	0.18
68,000–80,000	0.16	0.10
80,000–92,000	0.05	0.03

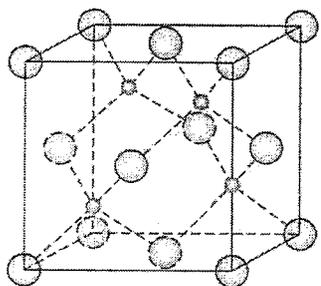
(Note: Degree of polymerization (DP) = M_n / m , M_n is the number-average molecular weight and m is the molecular weight of the monomer unit)

4. Match (2*4%)



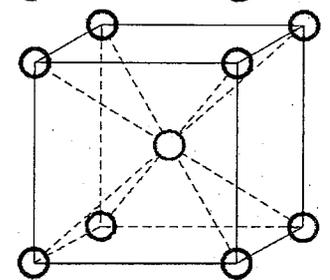
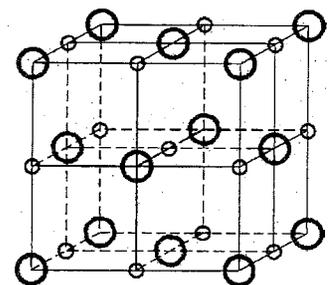
Cesium chloride

Zinc blende/ sphalerite



Rock salt

Fluorite



5. Identify the two crystal structure: Body-Centered Cubic (BCC) and Face-Centered Cubic (FCC). (r : radius of sphere and a : length of cube side) (2%*6)

Crystal Structure	Atom/ unit cell	The relation between r and a	Coordination number
BCC	(1)	(2)	(3)
FCC	(4)	(5)	(6)

6. On a tensile engineering stress – engineering strain plot of a ductile metal, indicate the following mechanical properties: (1) Yield strength, (2) Tensile strength, (3) Modulus of elasticity (5% × 3)
7. Briefly explain why HCP metals are typically more brittle than FCC and BCC metals. (8%)
8. Define the following terms: (1) strain hardening, (2) fracture toughness, and (3) heterogeneous nucleation. (5% × 3)
9. What is the driving force for grain growth of a polycrystalline material? (6%)
10. Plot a phase-transformation rate versus temperature curve. (6%)